



Pergamon

Tetrahedron Letters 40 (1999) 6581–6583

TETRAHEDRON
LETTERS

One-pot synthesis of a thioureido- β -cyclodextrin dimer

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Received 16 June 1999; accepted 4 July 1999

Abstract

The present work describes a one-pot synthesis of a thioureido- β -cyclodextrin isolated in good yield (32%) by the 'phosphinimine' approach. © 1999 Elsevier Science Ltd. All rights reserved.

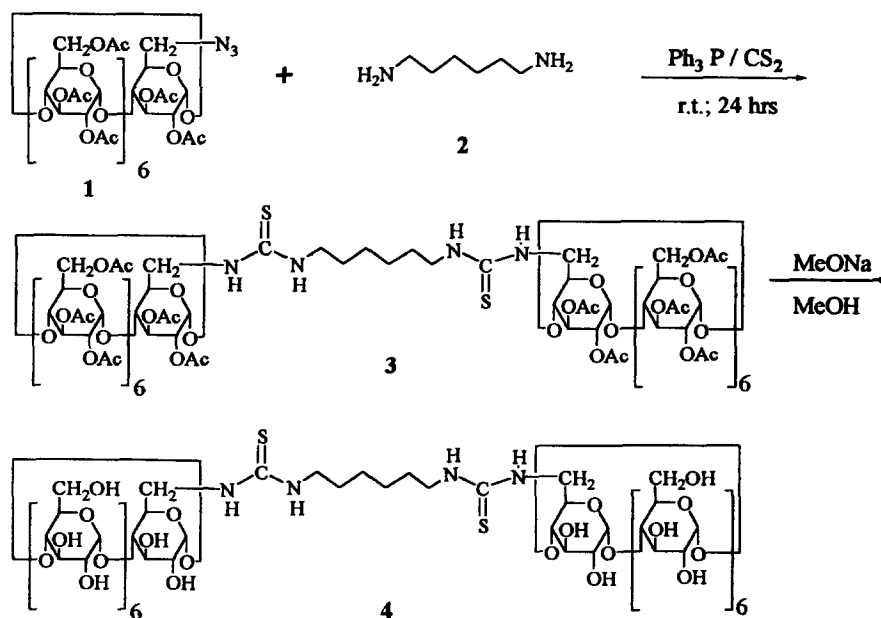
Keywords: β -cyclodextrin; Staudinger reaction; carbon disulphide; phosphinimine.

Pursuing developments of the 'phosphinimine' approach for direct access to abiotic receptors on the basis of cyclodextrins (Cds), we have recently reported studies leading to ureido Cds dimers^{1,2} and oligomers.³ Some of the published compounds have shown interesting biomimetic catalysis, and luminescent properties.³ The present work describes an extension of the 'phosphinimine' approach for general access to thioureido-Cd derivatives (Scheme 1). The one-pot condensation of the per-*O*-acetyl-6-monoazido-6-monodeoxy- β -cyclodextrin⁴ **1** with triphenyl phosphine (10 equiv.), hexamethylene diamine **2**, (0.6 equiv.) and CS₂ in anhydrous DMF as solvent, readily gives the acetylated *bis*-cyclodextrin ligand **3** in a good overall yield (32%) after 24 h of reaction and purification by chromatography on silica gel (CH₂Cl₂:MeOH, 98:2). The unprotected product **4** is obtained after a Zemplén deacetylation step (93%). The reaction could be also performed from the unprotected 6-monoazido-monodeoxy- β -Cd giving **4** under the same conditions in one step. The structures of new compounds **3–4** were fully analysed by IR, NMR and FABMS. The collected data are in agreement with the proposed structures.⁵ IR spectra of new compounds show characteristic ν (CS–NH) thiocarbonyl and amine ν (CS–NH) of thiourea functions. The mechanism is, probably, analogous to that determined in a previous work⁶ which suggests in situ formation of isothiocyanate-Cd intermediate during the reaction. Recently, literature has reported the reaction of nucleophilic amine with Cd-isothiocyanate applied to the synthesis of thioureido-Cds.⁷

Here, the 'phosphinimine' approach offers the advantage of a one-pot procedure leading to a wide variety of possible thio-derivatives without using hazardous reagents as isothiocyanates and thiophosgene.

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Scheme 1.

The above synthesised dimer 4 will be further explored for its potential biomimetic catalytic activity and metal complexation properties.

Acknowledgements

The service commun of NMR (Dr. P. Mutzenhart, Mrs. E. Eppiger), Université Henri Poincaré, Nancy-1 and the service commun of Mass Spectrometry (CNRS, Vernaison, France); CNRS, the Institut Nanceien de Chimie Moléculaire and Hungarian Scientific Research Fund (OTKA T23371) for financial support; Wacker Chimie S.A. (Lyon France) for their generous gift of β -cyclodextrin. The SAFAS S.A. Company (Monaco) for UV-vis facilities. We are indebted to Mrs. Nicole Marshall for correcting the manuscript.

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- Structures of all compounds were assigned by ^1H and ^{13}C NMR on a Bruker DRX-400 spectrometer, FTIR spectra were recorded on a Perkin-Elmer 1600 and mass spectra were recorded in FAB positive mode on a ZAB-SEQ mass spectrometer. The new compounds gave satisfactory analytical and spectroscopic data. **3**: Bis-[[hexakis(2,3,6-tri-*O*-acetyl)]-2,3-di-*O*-acetyl-cyclomaltoheptaosyl]-6-thioureido]-1,6-hexamethylenediamine obtained by condensation of **1** (0.5 mmol, 1.0 g, 1.2 equiv.), **2** (0.25 mmol, 0.029 g, 0.6 equiv.), carbon disulphide in large excess (40 mL) and triphenylphosphane (4.2 mmol, 1.09 g, 10 equiv.). White powder (0.33 g, 32%); TLC (CH_2Cl_2 :MeOH, 90:10): $R_f=0.3$; IR: 3626–3800 (N–H), 1732 (C=S); ^1H NMR (CDCl_3 , 25°C) δ (ppm): 5.40–5.20, 4.90–4.70, 4.20–3.95 (m, 21H (H_a2 , H_a3 , H_a5 , H_b2 , H_b3 , H_b5)); 5.18–4.95 (m, 14H (H_a1 , H_b1 , H_a4 , H_b4)); 4.60–4.40, 4.40–4.20 (m, 14H (H_a6 , H_b6)); 3.65–3.60

- (m, 2H, CH_2 , alkyl chain); 3.30 (m, 2H, CH_2 , alkyl chain); 2.30–2.00 (m, 60H, CH_3CO); 1.30 (m, 2H, CH_2 , alkyl chain); ^{13}C NMR ($CDCl_3$, 25°C) δ (ppm): 172–171 (CH_3CO); 170 (NH-C=S); 97 (C_{1a} , C_{1b}); 78–77 (C_{4a} , C_{4b}); 72–70 (C_{2a} , C_{3a} , C_{5a} , C_{2b} , C_{3b} , C_{5b}); 30 (CH_2 alkyl); 40 (CH_2 alkyl); 20 (CH_3CO); FABMS (NBA) : 3977 [$M-4[CH_3CO]+3H^+$] $^+$; Anal. calcd for $C_{172}H_{234}N_4O_{108}S_2$: C, 49.78; H, 5.68; N, 1.35; S, 1.54; found: C, 48.59; H, 5.66; N, 1.03; S, 1.45. 4: Bis-[cyclomaltoheptaosyl-6-thioureido]-1,6-hexamethylenediamine. White powder (0.182 g, 93%); TLC (dioxane:NH₃ 25%, 10:7 v/v): $R_f=0.8$; IR: 3708–3384 (N–H, O–H); 1731 (C=S); 1H NMR ($CDCl_3$, 25°C) δ (ppm): 5.40–5.20, 4.90–4.70, 4.20–3.95 (m, 21H (H_{a2} , H_{a3} , H_{a5} , H_{b2} , H_{b3} , H_{b5})); 5.18–4.95 (m, 14H (H_{a1} , H_{b1} , H_{a4} , H_{b4})); 4.60–4.40, 4.40–4.20 (m, 14H (H_{a6} , H_{b6})); 3.65–3.60 (m, 2H, CH_2 C1–C6 alkyl chain); 3.30 (m, 2H, CH_2 C2–C5 alkyl chain); 1.30 (m, 2H, CH_2 C3–C4 alkyl chain); ^{13}C NMR (D_2O , 25°C) δ (ppm): 170 (NH-C=S); 97.1 (C_{1a} , C_{1b}); 78–77 (C_{4a} , C_{4b}); 72–70 (C_{2a} , C_{3a} , C_{5a} , C_{2b} , C_{3b} , C_{5b}); 62 (C_{6a} , C_{6b}); 46 (CH_2 $C_{1,C6}$ alkyl); 40 (CH_2 C_2 , C_5 alkyl); 30 (CH_2 C_3,C_4 alkyl); FABMS (thioglycerol): 1379 [$M-[NH-CH_2-Cd]+2Na^+$] $^+$, 1199 [$Cd-CH_2-NH-CS]+Na^+$] $^+$, 1176 [$Cd-CH_2-NH-CS$] $^+$.
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